

# SYNTHESIS CHARACTERIZATION AND BIOLOGICAL STUDY OF ISOXAZOLINE

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A series of novel 4-tert-butyl-3-substituted-2-[5-(4- substituted phenyl)-4,5-dihydro-1,2oxazol-3-yl]phenolwere synthesized from different substituted 1-(3-tert-butyl-2-substituted-6hydroxyphenyl)-3-(4-substituted phenyl) prop-2-en-1-one.The structures of the compounds were elucidated by elemental and spectral (IR, <sup>1</sup>H NMR,) analysis. The synthesized compounds were checked for biological evaluation i.e. Antimicrobial, antifungal activity and Antioxidant Activity.

Keywords: chalcone, isoxazoline, biological evaluation, antimicrobial, antifungal study.

### **INTRODUCTION**

Isoxazoline are biologically active, synthetically useful, and important heterocycles having a wide role in medicinal chemistry. It is widely used as an antibacterial, anti-inflammatory, antifungal agent. Isoxazolines are also reported to possess good antimicrobial, analgesic, anti-inflammatory activity. In view of the biological activities, some isoxazoline derivatives and in continuation of our research work on synthesis of biologically active heterocyclic compounds, we investigated that the synthesis of some novel isoxazoline derivatives from

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chalconised derivatives is synthesized .The structure of all compounds was established on the basis of spectral and elemental analysis.

### Materials& Methods

### Step I –

Synthesis of 1-(3-tert-butyl-2-substituted-6-hydroxyphenyl) ethan-1-one (1-2)

### **General Method :**

In hot glacial acetic acid, fused zinc chloride was added and refluxed till solid was dissolved. Then powdered 4-tert-butyl-3-substituted phenol was added and refluxed for eight hours. The reaction mixture was cooled and then poured in acidulated water. The solid obtained was filtered, washed with water and recrystallized from rectified spirit to obtain titled compound. Thus 1-(3-tert-butyl-2-substituted-6-hydroxyphenyl) ethan-1-ones were synthesized

### Step II -

Synthesis of 1-(3-tert-butyl-2-substituted-6-hydroxyphenyl)-3-(4-substituted phenyl) prop-2en-1-one (3a-f)

In ethanol solvent, 1-(3-tert-butyl-2-substituted-6-hydroxyphenyl) ethan-1-one and aromatic aldehyde were added . To this mixture, dropwise added 10 % of KOH solution with constant stirring. The reaction mixture was kept overnight. Then this mixture was poured over HCland crushed ice. The product 1-(3-tert-butyl-2-substituted-6-hydroxyphenyl)-3-(4-substitutedphenyl)-prop-2-en-1-one was filtered and recrystallized from ethanol

**Step III** -Synthesis of Isoxazolinederivatives (4a-f)

An equimolar quantity of 1-(3-tert-butyl-2-substituted-6-hydroxyphenyl)-3-(4-substituted phenyl) prop-2-en-1-one, hydroxylamine hydrochloride andsodium acetate in ethanol solvent was refluxed for about 6hours. The mixture was concentrated by distilling out the solvent under reduced pressure and poured into ice water. The precipitate obtained was filtered, washed and recrystallized to get substituted isoxazolines.

### **Result & discussion:**

Scheme

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The present protocol for the synthesis of starting chalcones was done by using literature procedure, in which 4-substitudedaldehyde was mixture with 4-tert-butyl-3-substituted phenol in ethanol absolute and sodium hydroxide 10% to produce compound 3a-fThe different chalcones shows different product yield. As per literature, number of the procedure are available to carry out the reaction with different experiential condition but product utility of these product for different pathological and pharmaceutical in great importance.Finally 3a-f reacts with NH<sub>2</sub>OHto give substituted isoxazolines.

 Table1: Physical property of compounds

Comp	D1	DA	Mologular	MDo	% Viol	DE	% Nitrogen	
ounds	KI	R2	Formula	C NII °	d	K.F. Value	Found	Calcul
								ated
4a	Cl	Cl	$C_{19}H_{19}O_2NCl_2$	202	48%	0.41	3.83	3.85
4b	Cl	Br	C <sub>19</sub> H <sub>19</sub> O <sub>2</sub> NClBr	235	58%	0.47	3.41	3.43
4c	Cl	OH	C <sub>19</sub> H <sub>20</sub> O <sub>3</sub> NCl	199	49%	0.52	4.03	4.05
4d	Br	Cl	C <sub>19</sub> H <sub>19</sub> O <sub>2</sub> NClBr	192	42%	0.57	3.43	3.43
4e	Br	Br	$C_{19}H_{19}O_2NBr_2$	240	57%	0.62	3.08	3.09
4f	Br	OH	$C_{19}H_{20}O_3NBr$	175	42%	0.60	3.58	3.59

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### Spectral Analysis (Compound No. 4e):

**IR analysis (wave number in cm<sup>-1</sup>) 3100**-3000 (Ar-H stret.), 3100-300 (CH<sub>3</sub> stret.), 500-550 (C-Br stret.), 700-720 (C-Cl stret.), 3200-3300(-OH)

**NMR analysis** (δ ppm): 6.5-8 (Ar-H, 6H),8.93 (Ha),3.85-3.60 (Hb), 8.37(-OH,1H),1.5(-CH<sub>3</sub>, 9H)

## **BIOLOGICAL STYDY** Antimicrobial and antifungal activity Table 1: Antimicrobial activity

Sr. No.	Compounds	Escherichia coli	Pseudomonas aeruginosa	Staphylococcus Aureus	Bacillus subtilis
1	4a	14	16	17	18
2	4b	17	15	16	17
3	4c	16	16	14	18
4	4d	16	10	10	15
5	4e	11	14	18	10
6	4f	10	17	12	09

### Table2:Antifungal activity

Sr. No.	Compounds	Antifungal Activity		
		A. Niger	B. Albicans	
1	4a	18	17	
2	4b	12	16	
3	4c	17	18	
4	4d	15	12	
5	4e	13	11	
6	4f	14	07	

The antimicrobial and antifungal activity of all newly synthesized compounds was evaluated against gram-negative *Escherichia coli*, *Pseudomonas aeruginosa*, and gram-positive bacteria *Staphylococcusaureus*, *Bacillus subtilis*. The culture of each microbes species was incubated

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at 37  $^{\circ}$ c and the zone of inhibition on agar plates (diffusion method) was measured after 24 hrs. Most of these compounds were found active.

The antimicrobial and antifungal screening (Table-1 and 2) of above synthesized4-tert-butyl-3-substituted-2-[5-(4- substituted phenyl)-4,5-dihydro-1,2-oxazol-3-yl]phenol shows good activity against all microbes species. On the basis of screening data it was observed that these heterocyclic compounds can be easily used against treatment of disease caused by test microbes.

### Isoxazole derivatives :

### **Total phenolic content**

Total phenolic content was determined as described by Prior et al. [21]. Briefly, 500  $\mu$ g of compound in 100  $\mu$ L of methanol was mixed with 100  $\mu$ L of 1 N Folin–Ciocalteu reagent. Following incubation for 5 min, 200  $\mu$ L of 20% Na<sub>2</sub>CO<sub>3</sub> was added. Absorbance at 730 nm was measured in plate reader after 10 min and the concentration of phenolic compounds was calculated using standard curve of gallic acid (500–5000 ng; R<sup>2</sup>=0.967). The results were expressed as mg gallic acid equivalent (mg GAE) g<sup>-1</sup>.

### **Pyrimidine derivatives** :

Sr.No	Sample	µgGAE/mg
1	4- <i>tert</i> -butyl-3-chloro-2-[5-(4-chlorophenyl)-4,5- dihydro-1,2-oxazol-3-yl]phenol	19
2	4- <i>tert</i> -butyl-3-chloro-2-[5-(4-bromophenyl)-4,5- dihydro-1,2-oxazol-3-yl]phenol	09
3	4- <i>tert</i> -butyl-3-chloro-2-[5-(4-hydroxyphenyl)-4,5- dihydro-1,2-oxazol-3-yl]phenol	12
4	4- <i>tert</i> -butyl-3-bromo-2-[5-(4-chlorophenyl)-4,5- dihydro-1,2-oxazol-3-yl]phenol	15
5	4- <i>tert</i> -butyl-3-bromo-2-[5-(4-bromophenyl)-4,5- dihydro-1,2-oxazol-3-yl]phenol	17
6	4- <i>tert</i> -butyl-3-bromo-2-[5-(4-hydroxyphenyl)-4,5- dihydro-1,2-oxazol-3-yl]phenol	18

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### **REFERENCES**

- 1) Silva FM, Garden SJ, Pinto AC. The chemistry of isatin: A review from 1975 to 1999. J BrazChem. 2001;12:273–324.
- Pandeya SN, Sriram D, Nath G, DeClercq E. Synthesis, antibacterial, antifungal and anti HIV activities of Schiff and mannich bases derived from isatin derivatives and *N*-[4-(4'chlorophenyl) thiazol-2-yl] thiosemicarbazide. Eur J Pharm Sci. 1999;9:25– 31. [PubMed]
- 3) Sridhar KS, Pandeya SN, Bajpai KS, Manjula H. Synthesis antibacterial and antiviral activities of isatin derivatives. Indian Drugs. 1999;36:410–4.
- Mondal P, Banerjee M, Jana S, Bose A. Synthesis and evaluation of 1,3 di-Substituted Schiff, mannich bases and spiroisatin derivatives. J Young Pharm. 2010;2:169– 72. [PMC free article] [PubMed]
- 5) Nyati M, Rao SN, Srivastav KY, Verma LB. Microwave induced synthesis and antimicrobial activity of some 3-benzimidazolyl-5 aryl-2-isoxazolines. Indian J Hetro Chem. 2006;15:295–6.
- 6) Basawaraj R, Ahmed A, Khandre O, Sangapure S. Synthesis of some New pyrazolines and pyrimidines as potential antimicrobial agents. Indian J Hetro Chem. 2007;17:11–4.
- 7) Purohit SS. 6th ed. Agro Botanica Publishing Company; 2003. Microbiology: Fundamentals and applications; pp. 416–8.
- 8) Ghosh MN. Scientific. 2nd ed. Calcutta: Book Agency; 1984. Fundamentals of experimental pharmacology; pp. 153–8.
- 9) Dash GK, Suresh P, Kar DM, Ganpaty S, Panda SB. Evaluation of Evolvulusalsinoids Linn for anthelmintic and antimicrobial activities. J Nat Rem. 2002;2:182–5.
- 10) Kuppast IJ, Nayak V. Anthelmentic activity of fruits of *Cordiadichotoma*. Indian J Nat Prod. 2003;19:27–9.
- 11) Szewezuk VD, Mongelli ER, Pomilio AB. Antiparasitic activity of Meliaazadirach growing in Argentina. Mol Med Chem. 2003;1:54–5.
- 12) Shivkar YM, Kumar VL. Anthelmintic activity of latex of Calotropisprocera. Pharm Biol. 2003;41:263–5.

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